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## Key indicators

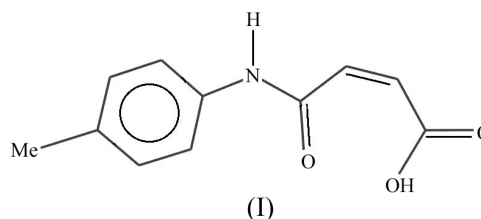
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.054  
 $wR$  factor = 0.177  
Data-to-parameter ratio = 12.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.***N*-(*p*-Tolyl)maleamic acid**

The title molecule,  $\text{C}_{11}\text{H}_{11}\text{NO}_3$ , is nearly planar, with the mean planes through the *p*-tolyl and maleamic acid groups inclined at an angle of  $5.45(3)^\circ$ . The glide-related molecules are linked by  $\text{N}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  intermolecular hydrogen bonds, forming infinite one-dimensional chains, which are assembled to form layers parallel to the *bc* plane.

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## Comment

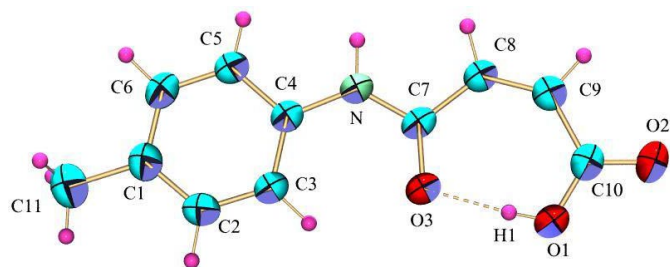
The structure of *N*-(*p*-tolyl)maleamic acid, (I), was reported previously by Prasad & Sinha (1978), using photographic X-ray diffraction data, with an  $R$  value of 0.16. The structure has now been refined using single-crystal X-ray diffraction data and the results are presented here.



The molecule of (I) is nearly planar, with C3 deviating by a maximum of  $0.103(3)$  Å. The mean planes through the *p*-tolyl and maleamic acid groups are inclined at an angle of  $5.45(3)^\circ$  (Table 1). The bond lengths and angles in the maleamic acid group agree with those reported for the structure of maleic acid (James & Williams, 1974). The carboxyl H atom is involved in an intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond with carbonyl atom O3. In the solid state, the glide-related molecules are linked by  $\text{N}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds (Table 2), forming infinite one-dimensional chains along the *c*-cell direction. These chains are arranged so as to form layers parallel to the *bc* plane, approximately at  $x = 1/4$  and  $3/4$ . The mean interlayer separation of  $3.406$  Å indicates significant interactions between the aryl and maleamic acid groups; the shortest interlayer contacts are  $\text{C6}\cdots\text{C10}(-x, 1-y, 1-z)$  of  $3.319(5)$  Å and  $\text{C4}\cdots\text{C7}(1-x, 1-y, 1-z)$  of  $3.351(4)$  Å.

## Experimental

The title compound was prepared by a solid-state reaction between *p*-phenylaniline and maleic anhydride by Professor R. P. Rastogi (Gorakhpur University) and his co-workers (private communication) and was recrystallized from methanol at room temperature.



**Figure 1**  
The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

#### Crystal data

$C_{11}H_{11}NO_3$	$D_x = 1.326 \text{ Mg m}^{-3}$
$M_r = 205.21$	Cu $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 25 reflections
$a = 6.769 (2) \text{ \AA}$	$\theta = 28.2\text{--}36.3^\circ$
$b = 12.109 (1) \text{ \AA}$	$\mu = 0.81 \text{ mm}^{-1}$
$c = 12.606 (1) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 95.73 (1)^\circ$	Block, colourless
$V = 1028.1 (3) \text{ \AA}^3$	$0.38 \times 0.31 \times 0.25 \text{ mm}$
$Z = 4$	

#### Data collection

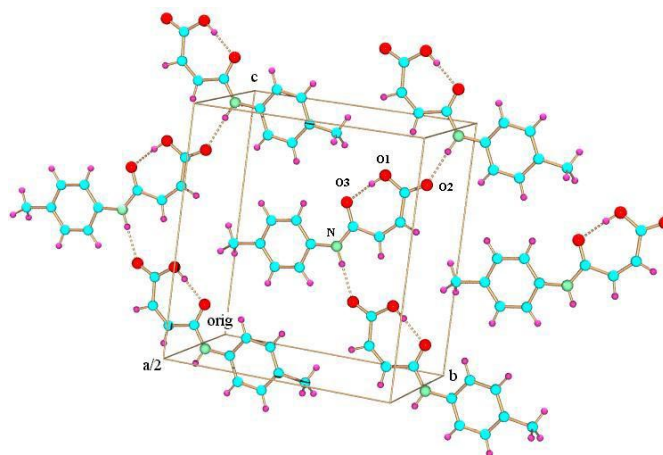
Enraf–Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.029$
$\omega$ - $2\theta$ scans	$\theta_{\text{max}} = 75.5^\circ$
Absorption correction: $\psi$ scan (North <i>et al.</i> , 1968)	$h = 0 \rightarrow 8$
$T_{\text{min}} = 0.733$ , $T_{\text{max}} = 0.817$	$k = 0 \rightarrow 15$
2123 measured reflections	$l = -15 \rightarrow 15$
2047 independent reflections	3 standard reflections
1200 reflections with $I > 2\sigma(I)$	frequency: 60 min
	intensity decay: none

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0899P)^2 + 0.2017P]$
$R[F^2 > 2\sigma(F^2)] = 0.054$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.178$	$(\Delta/\sigma)_{\text{max}} = 0.002$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
2047 reflections	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$
167 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.0101 (16)

**Table 1**  
Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

C4–N	1.412 (3)	C8–C9	1.337 (3)
C7–O3	1.240 (3)	C10–O2	1.214 (3)
C7–N	1.343 (3)	C10–O1	1.298 (3)
C7–C8	1.467 (4)		
C3–C4–N	124.3 (2)	C9–C8–C7	128.6 (2)
C5–C4–N	116.9 (2)	C8–C9–C10	132.6 (2)
N–C7–C8	113.5 (2)	C7–N–C4	128.9 (2)
O3–C7–N–C4	0.7 (5)	C3–C4–N–C7	−6.6 (5)
C8–C7–N–C4	−179.1 (3)	C5–C4–N–C7	174.3 (3)



**Figure 2**  
A view of the molecular layer, close to  $x = 1/4$ .

**Table 2**  
Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
N–H1N $\cdots$ O2 <sup>i</sup>	0.93 (3)	1.96 (3)	2.885 (3)	177 (2)
O1–H1O $\cdots$ O3	0.82	1.68	2.495 (2)	173
C3–H3 $\cdots$ O3	0.98 (3)	2.26 (2)	2.866 (3)	119 (2)
C8–H8 $\cdots$ O1 <sup>i</sup>	0.97 (3)	2.55 (2)	3.498 (3)	164 (2)

Symmetry code: (i)  $x, \frac{3}{2} - y, z - \frac{1}{2}$ .

All H atoms apart from the methyl and hydroxyl H atoms were refined isotropically.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *MolEN* (Fair, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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